VICINAL ALKYNYLATION-ALKYLATION VIA A TANDEM MICHAEL ADDITION TRANSITION METAL CATALYZED ALKYLATION REACTION

Barry M. Trost, Chao-Jun Li Department of Chemistry Stanford University Stanford, California 94305-5080

Summary: A protocol whereby addition of alkynylalanes to alkylidenemalonates is followed by palladium catalyzed allylation permits vicinal β -alkynylation α -allylation in a single operation.

In conjunction with our interest in the synthesis of the potent nonadride herbicide cornexistin 1,¹ we became interested in the vicinal alkylation illustrated in eq. 1. While sequential conjugate addition of alkyl and

alkenyl groups via organocuprate chemistry followed by alkylation is well documented,² such reactions initiated by conjugate addition of alkynes have normally not been successful. Conjugate additions of alkynes require alkynylaluminum^{3,4} and alkynylborane⁵ reagents.⁶ The low reactivity of the resulting enolate makes further alkylation reactions improbable. On the other hand, the good reactivity of palladium catalyzed alkylations with highly stabilized nucleophiles⁷ led us to examine the feasibility of the sequence of eq. 1 wherein the allylation is catalyzed by palladium.

We explored dimethyl ethylidenemalonate (2) as a test substrate (eq. 2). Diethylphenylethynylalane (3), generated in normal fashion by lithiation with n-butyllithium followed by transmetalation with diethylaluminum chloride, reacted with 2 at room temperature in a 5:3 hexane-ether mixture to generate presumably the aluminum

enolate 4. Addition of an allyl acetate and a Pd(0) complex failed to give any allylation product. On the other hand, replacement of the hexane-ether with THF followed by allyl acetate and addition of the resultant solution to

a THF solution of a Pd(0) catalyst (2 mol%) generated from (dba)₃Pd₂•CHCl₃ (5) and 9 mol% TTP (triphenylphosphine) gave the desired vicinal disubstituted product 6 in 78% yield after heating at reflux for 5 h.

Using the acetylide from the silyl ether of propargyl alcohol under otherwise identical conditions led to smooth reation to give the vicinal disubstituted product 78 (eq. 3). On the other hand, changing the allyl partner

required changing the solvent and/or ligand. For example, adding butadiene monoepoxide to 4 as described above led to no alkylation. However, using DMSO instead of THF with all other factors being the same produced the desired double alkylation product 88 (eq. 4, path a). With 3-acetoxy-3-methyl-1-butene, both the solvent and

ligand required modification. The doubly substituted product 98 was produced using DMSO as solvent and triphenyl phosphite as ligand (eq. 4, path b). Interestingly, neither TPP nor triisopropyl phosphite served as ligands for this reaction.

Substituents on the alkylidenemalonate did not have much of an effect. Increasing the steric hindrance by switching from methyl to isopropyl as in 10 or introducing a conjugating substituent like phenyl as in 11 did not affect the reaction. In both cases, the original protocol generated the desired products 128 and 13.8

The requirement for as non-coordinating a solvent as possible for the addition of the alkynylalane and a more polar solvent for the allylic alkylation necessitated the change of solvent in the above procedure. A

modification of the conditions for the first stage obviated the need for this change in solvent. Forming the alane 3 in 6:1 hexane-THF still permitted the conjugate addition. Simply diluting the resultant reaction mixture with an additional 20 parts of THF then allowed the palladium catalyzed allylic alkylation to proceed. For these experiments, π -allylpalladium chloride dimer was employed as the palladium(0) source. In this way, 6 (eq. 2) and 12 (eq. 5) were obtained in 57% and 65% yields respectively.

Experimental Procedure. A solution of n-butyllithium in hexane (0.8 mL, 1.5 M, 1.2 mmol) followed after 15 min by 0.5 mL of THF and a solution of dischylaluminum chloride in hexane (0.44 mL, 2.5 M, 1.1 mmol) was added to phenylacetylene (102 mg, 1 mmol) in 2 mL of hexane at 0°. After stirring 30 min, dimethyl ethylidenemalonate (158 mg, 1 mmol) was added and the reaction stirred 3 h at room temperature. After sequential addition of 10 mL of THF, 150 mg (1.5 mmol) of allyl acetate, 24 mg (0.09 mmol) of TPP, and 3.7 mg (0.01 mmol) of π -allylpalladium chloride, the reaction was heated at 75-80° for 10 h, cooled, diluted with hexane, filtered and concentrated. Chromatography of the residue (25:1 hexane-ethyl acetate) gave 171 mg (57%) of 6.

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References:

- 1. Nakajima, M.; Itoi, K.; Takamatsu, Y.; Sato, S.; Furukawa, Y.; Furuya, K.; Honma, T.; Kadotani, J.; Kozasa, M. J. Antibiotics 1991, 44, 1065.
- 2. Kozlowski, J.A. in "Comprehensive Organic Synthesis," Trost, B.M.; Fleming, I.; Semmelhack, M.F. eds, Pergamon Press, 1991, Vol. 4, pp 169-198; Hulce, M.; Chapdelaine, M.J. idid. pp 237-268.
- Hooz, J.; Layton, R.B. J. Am. Chem. Soc. 1971, 93, 7320; Pappo, R.; Collins, P.W. Tetrahedron Lett. 1972, 2627; Collins, P.W.; Dajan, E.Z.; Bruhn, M.S.; Brown, C.H.; Palmer, J.R.; Pappo, R. Tetrahedron Lett. 1975, 4217; Hashimoto, S.I.; Shinoda, T.; Ikegami, S. Tetrahedron Lett. 1986, 27, 2885.
- For a Ni(0) catalyzed version, see Hansen, R.T.; Carr, D.B.; Schwartz, J. J. Am. Chem. Soc. 1978, 100, 2244; Schwartz, J.; Carr, D.B.; Hansen, R.T.; Dayrit, F.M. J. Org. Chem. 1980, 45, 3053.
- Bruhn, M.; Brown, C.H.; Collins, P.W.; Palmer, J.R.; Dajani, E.Z.; Pappo, R. Tetrahedron Lett. 1976,
 235; Sinclair, J.A.; Molander, G.A.; Brown, H.C. J. Am. Chem. Soc. 1977, 99, 954; Fujishima, H.;
 Takada, E.I.; Hara, S.; Suzuki, A. Chem. Lett. 1992, 695.
- For a Rh(I) catalyzed addition of terminal acetylenes to conjugated carbonyl compounds, see Nikishin, G.I.; Kovalev, I.P. *Tetrahedron Lett.* 1990, 31, 7063.
- Trost, B.M. Tetrahedron 1977, 33, 2615; Trost, B.M.; Accounts Chem. Res. 1980, 13; 385; Trost, B.M.; Verhoeven, T.R. Compr. Organomet. Chem. 1982, 8, 799; Tsuji, J. "Organic Synthesis with Palladium Compounds," Springer, Berlin, 1980; Tsuji, J. Tetrahedron 1986, 42, 4361; Tsuji, J.; Minami, I. Accounts Chem. Res. 1987, 20, 140; Trost, B.M. Angew. Chem. Int. Ed. Engl. 1989, 28, 1173; Godleski, S.A. in "Comprehensive Organic Synthesis" Trost, B.M.; Fleming, I.; Semmelhack, M.F. 1991, Vol. 4, pp 585-662.
- 8. Characterization of the products:

6: 1 H NMR (CDCl₃): δ 1.4 (d, J = 7.0Hz, 3H), 2.88 (dd, J = 1.1, 7.3Hz, 2H), 3.40 (q, J = 7.0Hz, 1H), 3.76 (s, 3H), 3.79 (s, 3H), 5.05-5.25 (m, 2H), 5.82 (m, 1H), 7.25-7.45 (m, 5H). 13 C NMR (CDCl₃): δ 17.51, 31.21, 38.59, 52.28, 52.35, 61.37, 83.26, 89.73, 118.96, 123.42, 127.87, 128.19, 131.57, 132.92, 169.93, 170.28. IR (neat): 2200, 1736, 1630, 1605, 1580, 1485, 1435 cm. $^{-1}$ HRMS: Calc'd for C₁₈H₂₀O₄: 300.1362. Found: 300.1343.

7: ¹H NMR (CDCl₃): δ 0.11 (s, 6H), 0.90 (s, 9H), 1.29 (d, J = 7.1Hz, 3H), 2.80 (d, J = 7.2Hz, 2H), 3.20 (q, J = 6.9Hz, 1H), 3.72 (s, 3H), 3.75 (s, 3H), 4.30 (d, J = 2.1Hz, 2H), 5.10 (m, 2H), 5.8 (m, 1H). ¹³C NMR (CDCl₃): δ -5.44, 17.26, 18.03, 25.60, 30.54, 38.37, 51.73, 52.09, 52.17, 61.04, 81.74, 84.68, 119.03, 133.09, 170.21, 170.60. IR (neat): 2270, 1735, 1640, 1463, 1435 cm. ⁻¹ HRMS: Calc'd for C₁\$H₂₃O₅ (M+-C₄H₉): 311.1315. Found: 311.1307.

8: ¹H NMR (CDCl₃): δ 1.38 (d, J = 6.9Hz, 3H), 1.6 (br, 1H), 2.88 (m, 2H), 3.40 (q, J = 6.9Hz, 1H), 3.76 (s, 3H), 3.78 (s, 3H), 4.08 (d, J = 4.6Hz, 2H), 5.75 (m, 2H), 7.2-7.5 (m, 5H). ¹³C NMR (CDCl₃): δ 17.25, 30.98, 36.74, 52.25, 52.32, 61.34, 63.33, 83.41, 89.72, 126.84, 128.12, 128.40, 128.56, 131.74, 133.76, 170.20, 170.60. IR (neat): 3400, 2250, 1731, 1600, 1435 cm. ⁻¹ HRMS: Calc'd for C₁₉H₂₀O₄ (M⁺-H₂O): 312.1362. Found: 312.1349.

9: ¹H NMR (CDCl₃): δ 1.38 (d, J = 7.0Hz, 3H), 1.65 (s, 3H), 1.70 (s, 3H), 2.78 (dd, J = 7.1, 14.8Hz, 1H), 2.92 (dd, J = 7.9, 14.7Hz, 1H), 3.41 (q, J = 7.0Hz, 1H), 3.74 (s, 3H), 3.77 (s, 3H), 5.1 (m, 1H), 7.2-7.5 (m, 5H). ¹³C NMR (CDCl₃): δ 17.33, 17.67, 25.89, 30.73, 32.62, 52.09, 52.16, 61.21, 83.04, 90.21, 118.20, 123.71, 127.94, 128.32, 131.75, 135.79, 170.60, 170.92. IR (neat): 1735, 1580, 1435 cm.-¹ HRMS: Calc'd for C₂₀H₂₄O₄: 328.1675. Found: 328.1654.

12: 1 H NMR (CDCl₃): δ 0.93 (d, J = 6.6Hz, 3H), 1.12 (d, J = 7Hz, 3H), 2.15 (m, 1H), 2.88 (dd, J = 7.5, 13Hz, 1H), 3.05 (dd, J = 7.4, 14Hz, 1H), 3.30 (d, J = 2.4Hz, 1H), 3.76 (s, 6H), 5.15 (m, 2H), 5.80 (m, 1H), 7.3-7.5 (m, 5H). 13 C NMR (CDCl₃): δ 18.64, 23.80, 28.13, 39.08, 43.04, 52.17, 52.24, 60.68, 85.62, 86.63, 119.26, 123.88, 127.89, 128.32, 131.76, 132.81, 170.47, 171.16. IR (neat): 2200, 1751, 1625, 1600, 1490, 1436 cm. $^{-1}$ HRMS: Calc'd for $C_{20}H_{24}O_4$: 328.1675. Found: 328.1674.

13: 1 H NMR (CDCl₃): δ 2.57 (dd, J = 7.3, 14.3Hz, 1H), 3.0 (dd, J = 7.1, 14.3Hz, 1H), 3.64 (s, 3H), 3.78 (s, 3H), 4.73 (s, 1H), 5.15 (m, 2H), 5.85 (m, 1H), 7.25-7.5 (m, 10H). 13 C NMR (CDCl₃): δ 38.09, 42.44, 52.00, 52.16, 63.16, 84.88, 88.32, 118.85, 123.44, 127.90, 128.14, 128.29, 128.43, 129.75, 131.71, 133.24, 136.95, 169.79, 169.94. IR (neat): 1741, 1625, 1600, 1491, 1455, 1435 cm. $^{-1}$ HRMS: Calc'd for C₂₃H₂₂O₄: 362.1518. Found: 362.1545.

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